A Nanoscale Plasmonic Reactor: Light-Driven Synthesis of Individual Core@Shell Nanoparticles

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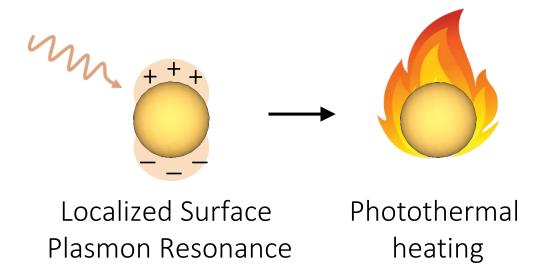
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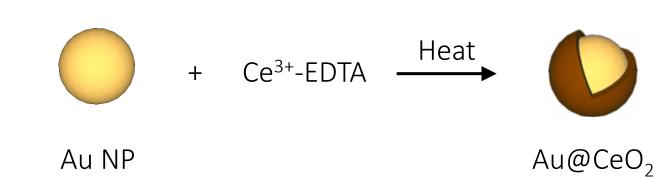
1. Motivation: Non-radiative decay of plasmons can drive chemistry



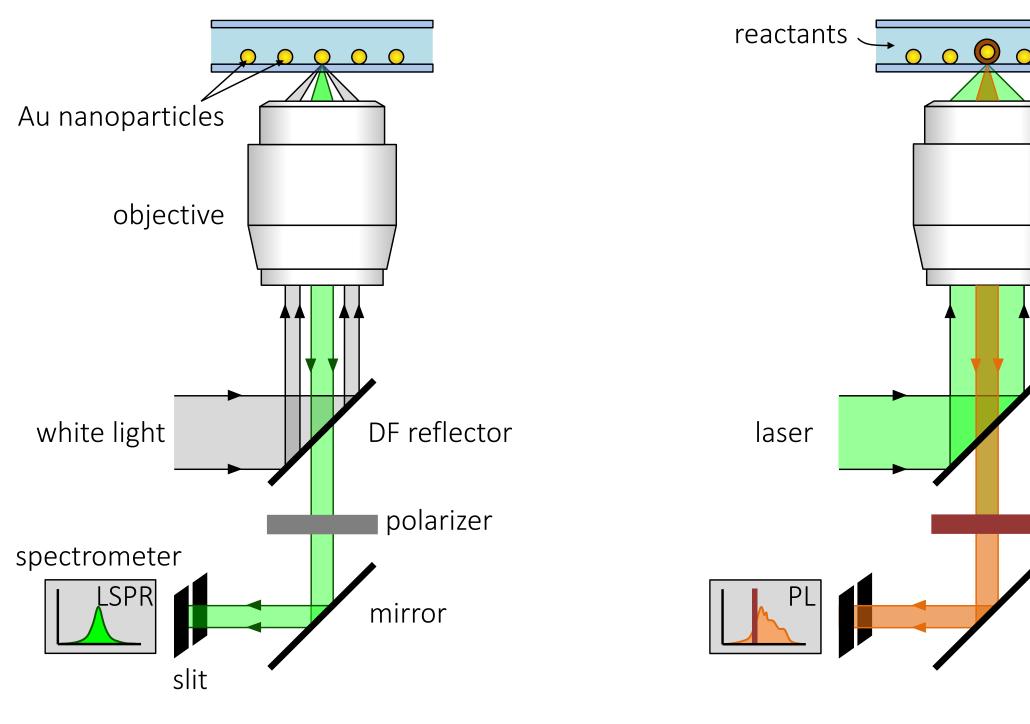
Nanostructures made of noble metals such as Au, Ag and Cu support localized surface plasmon resonances (LSPRs), which are collective oscillations of their free electrons, driven by light. These surface plasmons decay by locally heating the nanoparticles, which can drive chemical reactions.²

Here, we demonstrate how localized photothermal effects can be used to produce spatially-confined nanoreactors by activating, controlling, and spectroscopically following the growth of individual metal@semiconductor core@shell nanoparticles.

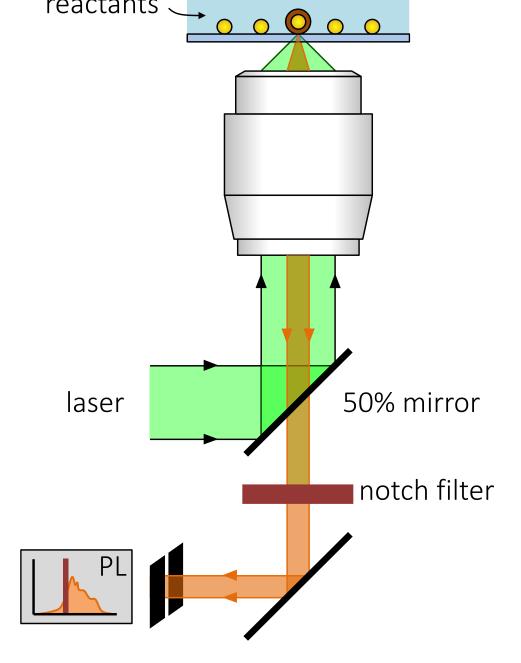
2. Demonstrating the photothermal activation and nanoscale control of chemical reactions, using a core@shell synthesis



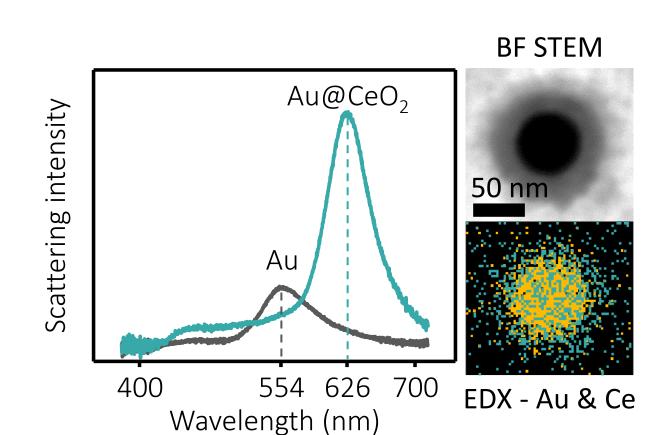
This temperature-activated Au@CeO2 core@shell nanoparticle (NP) synthesis occurs at higher temperatures such as 90 °C in ensemble conditions. We perform this synthesis under photothermal heating of 'individual' Au NPs.

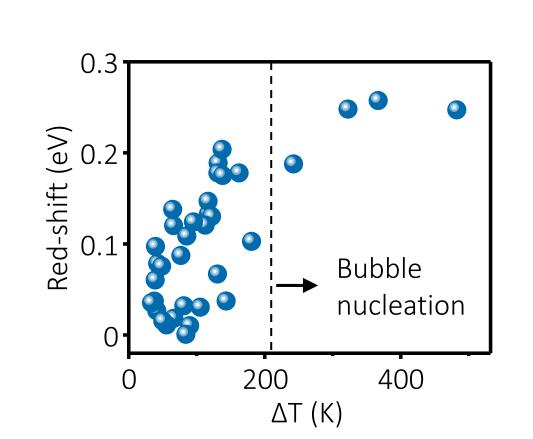


Dark-field single-particle scattering spectroscopy

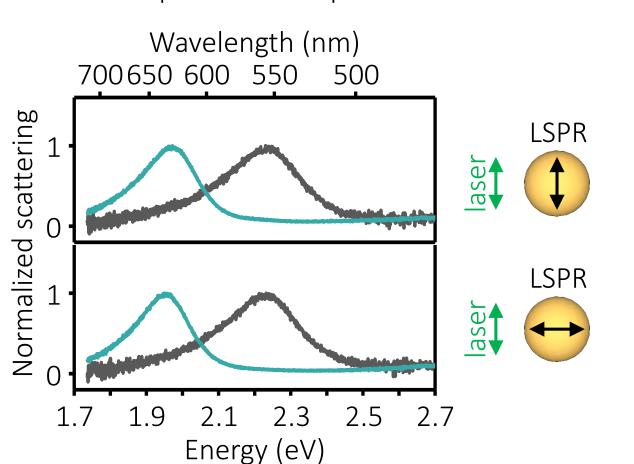


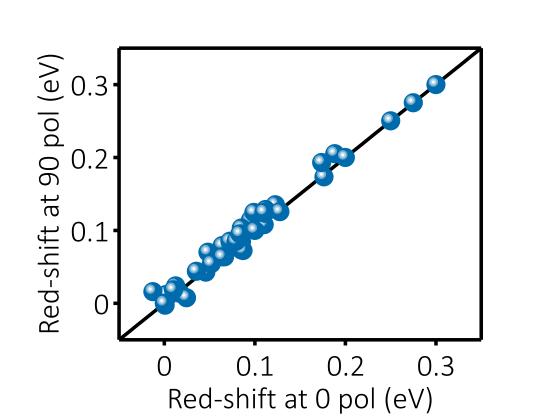
Single-particle irradiation & *In situ* photoluminescence spectroscopy





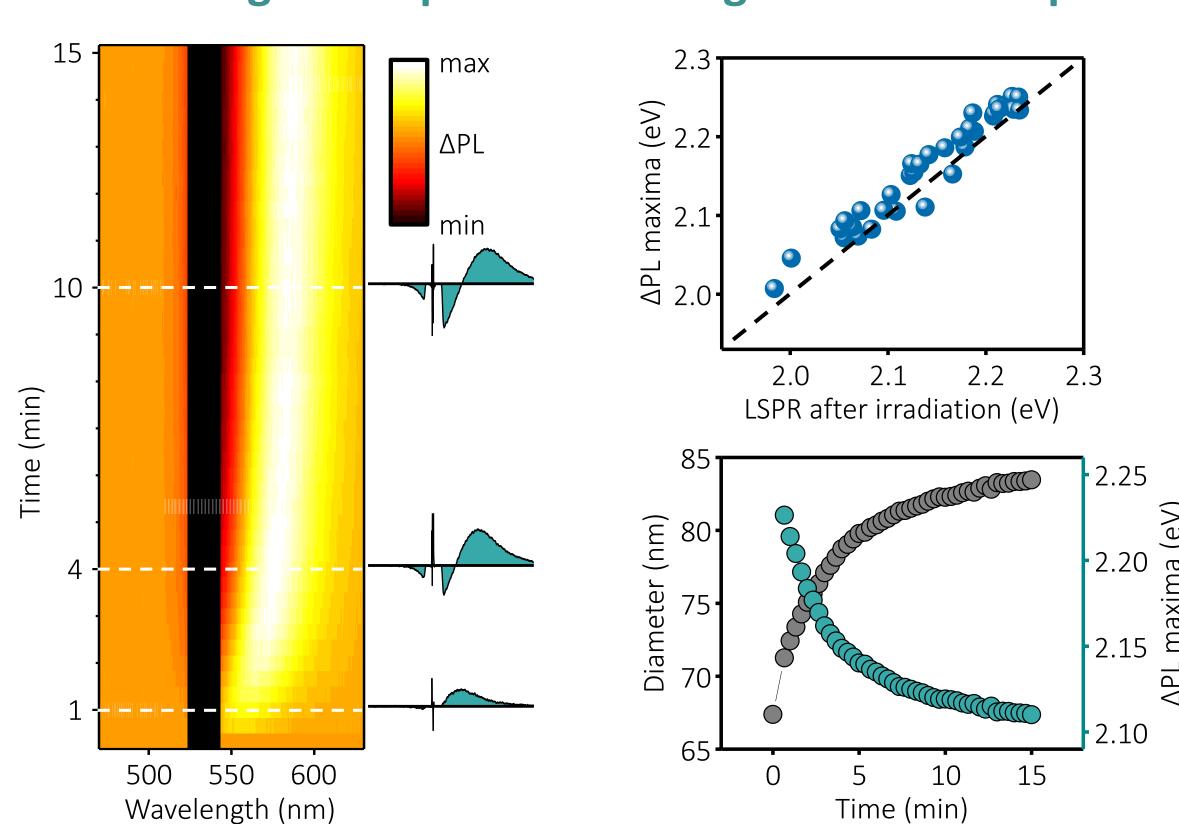
Under irradiation, we observe a red-shift in the LSPR of Au NP due to the growth of CeO₂ shell. Electron microscopy and EDX maps confirm the formation of Au@CeO2 core@shell NPs. The photothermal shell growth scales with the nanoparticle temperature.





Photothermally-grown ceria shell is isotropic in nature, as confirmed by polarization-dependent scattering spectra on several individual NPs.

3. In situ tracking of the photothermal growth of nanoparticles

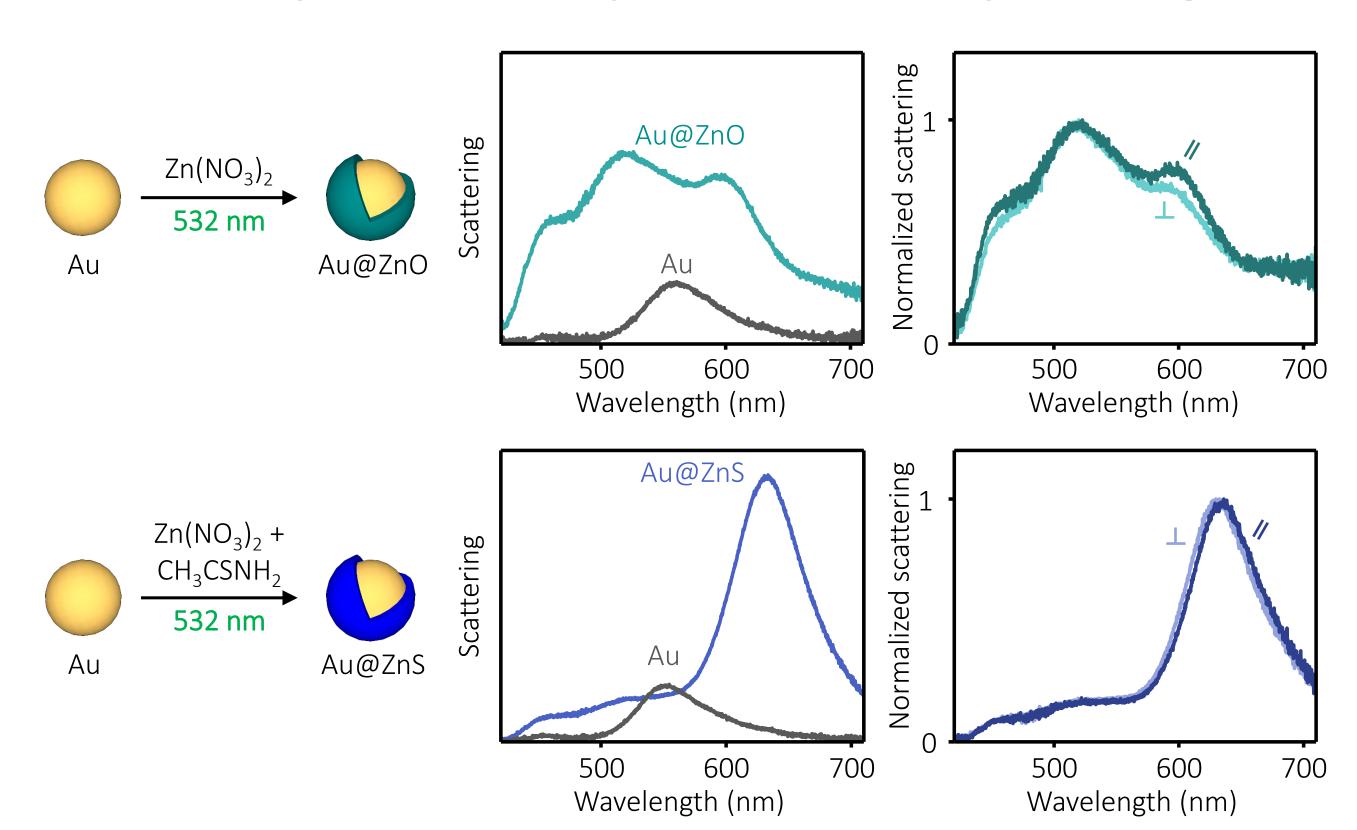


Real-time tracking of the photoluminescence of the growing core@shell nanoparticle is useful to study their growth kinetics.

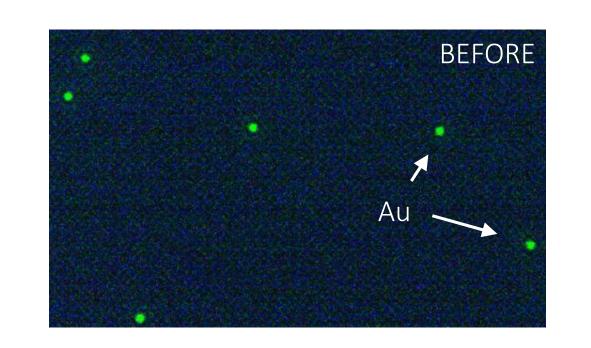
5. Conclusion & Outlook

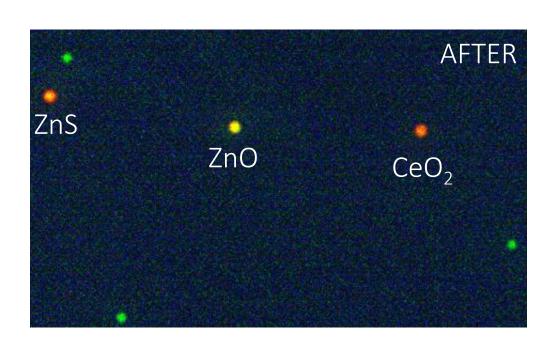
- Localized temperature gradients produced by plasmonic heating can be exploited to drive spatially confined chemical reactions.
- Combining our technique with automated particle centering algorithms, one can envisage fast printing of hierarchical nanoparticles with advanced functionalities over large areas.

4. Versatility of the technique and substrate patterning



We demonstrate the growth of metal oxides and metal sulfides using photothermal nanoreactors.





References

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Acknowledgements

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